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solutions this activity will explore whether or not there is a limit to how much of one substance can dissolve in another model 1 saturated and unsaturated solutions all beakers

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add a small amount of solute of known mass if it dissolves it is unsaturated if solid settles to the bottom filter the solution and mass the remaining solid if the mass is the same then the solution was saturated if the mass is lower than it is an unsaturated solution model 2 solute dissolved vs solute added

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model 1 saturated and unsaturated solutions unsaturated solutions all beakers contain 100 g of water all beakers are kept at 20°C all solutions are stirred for 2 hours solute is the same substance in all beakers number of dissolved particles number of solid particles beaker c 0.00036 g of solute added beaker a i g of solute added

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different solutes such as salt sugar or minerals dissolve to very different extents in water and other solvents in this activity you will learn how to quantify the amount of solute that is dissolved in a saturated solution

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chem 116 pogil worksheet week 4 properties of solutions key questions 1 identify the principal type of solute solvent interaction that is responsible for forming the following solutions a KNO<sub>3</sub> in water b Br<sub>2</sub> in benzene c 6H<sub>6</sub> c glycerol ch 22 OH CH OH CH OH in water d HCl in acetonitrile ch 3 CN HCl does not form ions in

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to prepare that volume of a 3.00 M acetic acid solution from the 15.0 M standard solution determine the dilution factor used to prepare any of these solutions check your group's answers by comparing with other groups resolve any differences 6 suppose shelby's supervisor said to make 250 mL of 3.00 M acetic acid solution how would you

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a saturated solution is a solution that contains the maximum amount of solute that is capable of dissolving an unsaturated solution is a solution that contains less than the maximum amount of solute that is capable of being dissolved solution equilibrium exists when the rate of dissolution equals the rate of recrystallization

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this activity will explore whether or not there is a limit to how much of one substance can dissolve in another model 1 saturated and unsaturated solutions all beakers contain 10.0 g of water all beakers are kept at 20 °C all solutions are stirred for 2 hours solute is the same substance in all beakers

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a saturated solution is a solution that contains the maximum amount of solute that is capable of dissolving an unsaturated solution is a solution that contains less than the maximum amount of solute that is capable of being dissolved solution equilibrium exists when the rate of dissolution equals the rate of recrystallization

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solution click the card to flip mixture that has the same composition color and density throughout homogeneous mixture click the card to flip 1 15 flashcards learn test match 1 4 study with quizlet and memorize flashcards containing terms like solution solute solvent and more

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this interactive lecture demonstration guide has the following goals students will be able to 1 compare and describe saturated and unsaturated solutions at the particle level and in terms of macroscopic observations 2 explain how and whether changes in solute amount and changes in volume affect the concentration of unsaturated and

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