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some students investigated the rate of decomposition of hydrogen peroxide H_2O_2 the equation for the reaction is $2 \text{H}_2\text{O}_2 \text{ aq} \rightarrow 2 \text{H}_2\text{O} \text{ l} + \text{O}_2 \text{ g}$ the catalyst for the reaction is manganese dioxide 1 describe a test to identify the gas produced in the reaction give the result of the test

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the total mark for this paper is 80 the marks for each question are shown in brackets use this as a guide as to how much time to spend on each question advice read each question carefully before you start to answer it try to answer every question check your answers if you have time at the end

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general instructions to the teacher conducting the speaking assessment the assessment is made up of three tasks the tasks must be conducted in the following order task 1 role play task 2 picture based task task 3 conversation

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