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step 1 of 1 the vapor pressure lowering is governed by raoult s law it states that the vapor pressure of the solvent above the solution is equal to vapor pressure of the pure solvent at that temperature multiplied by the mole fraction of the solvent present the expression is given as follows

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solution length of the room  $l = 5 \text{ m}$  breadth of the room  $b = 4 \text{ m}$  height of the room  $h = 3 \text{ m}$  the room is like a cuboid whose four walls lateral surface and ceiling are to be white washed area for white washing lateral surface area  $2 lb + 2 lh$  area of the ceiling  $l \times b$

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