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base your answers to questions 66 through 69 on the information below before atomic numbers were known mendeleev developed a classification system for the 63 elements known in 1872 using oxide formulas and atomic masses he used an r in the oxide formulas to represent any element in each group

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the document contains the results of several experiments 1 the solubility of potassium iodide ki was determined at different temperatures showing that more ki dissolves at higher temperatures 2 common laboratory techniques like filtration and evaporation were used to separate sodium chloride from a mixture of salts

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each atom now has eight electrons that is a stable octet of electrons around it in part c iii candidates were required to write the chemical formula of the compound formed between p and y and suggest two properties of this compound a number of candidates showed the correct bonding diagram

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