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## ***solution definition examples britannica***

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learn what a solution is in chemistry how it is formed and classified and what are its applications and properties explore the differences between liquid gas and solid solutions and the concepts of solubility electrolytes and alloys

## **solution chemistry wikipedia**

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in chemistry a solution is a special type of homogeneous mixture composed of two or more substances in such a mixture a solute is a substance dissolved in another substance known as a solvent

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a solution is a homogeneous mixture of two or more pure substances the substance that is in a large amount in the solution is called the solvent the substance that is in smaller amounts in a solution is called the solute for example the air is a solution in which nitrogen is the solvent and water is the solvent in seawater and body fluids

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learn what a solution is and how to identify its components types and properties a solution is a homogeneous mixture of two or more substances in which one is dissolved in another

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we call the major component the solvent and the minor component s the solute s the most familiar solutions are aqueous solutions in which water is the solvent for example in a solution of the sugar glucose in water glucose molecules are the solute and water molecules are the solvent

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a solution is a homogeneous mixture of two or more substances in one phase learn about the characteristics types and examples of solutions in chemistry

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a solution is a homogeneous mixture of one or more solutes dissolved in a solvent learn about different kinds of solutions and their microscopic behavior with examples and figures

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solutions are composed of particles the size of an ion or small molecule  $0.1 - 2.0 \text{ nm}$  the examples provided above are all considered solutions air is a solution of small gas molecules simple syrup is a solution of sucrose in water and seawater is a solution of ions and water

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introduction to pH prepare to mix it up in this unit on solutions acids and bases after differentiating between types of mixtures we'll dive into solutions learning about solvation and solubility then we'll learn about acids and bases both corrosive and caustic get ready to dissolve your doubts and explore the mysteries of solubility

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learn what a solution is how to classify it based on the states of matter of solute and solvent

and how to measure its concentration explore the properties and examples of solutions with byju s videos and worksheets

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a solution is what occurs when two chemicals are mixed referred to as a solvent and a solute the combination of solute and solvent together is a solution these are often one substance dissolved in another such as dissolving salt in water although a solution can be made from any state of matter combining a solute and a solvent makes a solution

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learn what a solution is in chemistry how to identify its parts types properties and examples a solution is a homogeneous mixture of two or more substances that can exist in any phase of matter

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learn what a solution is and how it is formed by solvent and solute molecules explore different types of solutions such as polar nonpolar acidic and basic and see examples from biology and chemistry

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a solution in science is a homogeneous mixture made up of two or more components and can be a solid liquid or gas common solutions include steel air and salt water what makes up a solution in biology for consideration as a solution a mixture must have at least two components and be homogeneous

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solutions and alloys are types of homogeneous mixtures 1 bubble solution you can make a simple bubble solution for blowing bubbles by combining liquid dishwashing soap and water but chemistry holds the secrets to making better bubble solutions that create larger or longer lasting bubbles

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a scientific solution is defined as two or more substances in a homogenous mixture discover the parts of a solution and see examples of the three

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introduction a solution is a mixture of two or more substances that stays evenly mixed substances that are combined to form a solution do not change into new substances some examples of solutions include seawater gasoline glass steel and air mixing and separating

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solutions are a homogeneous mixture of two or more substances it has homogeneity at the particle level usually people think of it as some liquid with either a solid or a liquid or a gas dissolved in it however this is not entirely true we can also have solid solutions like alloys for example air it is a mixture of gas in gas

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solutions play a very important role in many biological laboratory and industrial applications of chemistry of particular importance are solutions involving substances dissolved in water or

aqueous solutions solutions represent equilibrium systems and the lessons learned in the last chapter will be of particular importance again

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dissolving bbc bitesize part of chemistry pure and impure substances remove from my bitesize key points a solution is made when a solute dissolves into a solvent if a substance can

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