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express your answer to the correct number of significant figures one milligram is equal to 0.001 grams drag each number to the correct location sort the numbers based on the number of significant figures they have

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number of atoms of 1st element in chemical formula x atomic mass of 1st element number of atoms of 2nd element in chemical formula x atomic mass of 2nd element study with quizlet and memorize flashcards containing terms like solve the following problems 1 calculate the molar mass for MgCl_2

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the chemical formula for aspirin is $C_9H_8O_4$ calculate the mass percent composition of aspirin

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using moles avogadro s number 1.267×10^3 mol of h and 2.67×10^3 mol of cl 5.34×10^3 mol of h 2.67×10^3 mol of s and 10.68×10^3 mol of o 2.67×10^3 mol of al 8.01×10^3 mol of n and 24.03×10^3 mol of o

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preview this quiz on quizizz what is the empirical formula of a substance that is 53 5 c 15 5 h and 31 1 n by weight

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the following quiz deals with the physical properties of chemical substances it will test your abilities on mixtures and pure substances the quiz is supposed to test your knowledge and prepare you for more advanced ones good

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this chapter will describe how to symbolize chemical reactions using chemical equations how to classify some common chemical reactions by identifying patterns of reactivity and how to determine the quantitative relations between the amounts of substances involved in chemical reactions that is the reaction stoichiometry

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preview this quiz on quizizz if you are solving for mass and you know the number of moles which conversion should you use

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