

# Free ebook Nuclear reactions study guide (Read Only)

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the general types of chemical reactions are combination decomposition single replacement double replacement and combustion reactions net ionic equations show only those particles involved in a double replacement reaction generally takes place between two ionic compounds in aqueous

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steps to complete a reaction be able to fill in the missing products or reactants for chemical reactions identify the type of reaction determine the reaction mechanism determine if the reaction will occur

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identify the type of reaction write a balanced equation incl states 8 aqueous solutions of ammonium chloride and lead ii nitrate produce lead ii chloride precipitate and aqueous ammonium nitrate 9 solid carbon disulfide burns in oxygen to yield carbon dioxide and sulfur dioxide gases 10

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recognizing single displacement reaction are oxidation reduction reaction and identifying which elements was oxidized and which one was reduced recognizing that acid base neutralization reactions produce water and an ionic compound in textbook do these problems for practice for the test

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