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procedure 1 all half reactions on the standard reduction potentials table are compared to hydrogen  $E^0 = 0.2$  all half reactions can be read in reverse as oxidations in which case the sign of the voltage  $E^0$  is changed 3 the net voltage is the sum of the voltages of the oxidation half reactions and the reduction half reactions see chart

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answer the questions below based on the information above and on your knowledge of chemistry 1 what is the charge on a proton 2 what is the charge on an electron 3 why do an atom s electrons revolve around its protons instead of drifting away 4 why are the electrons of one atom attracted to the protons of another 5

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answer the questions below based on your reading above and on your knowledge of chemistry 1 dry ice is solid carbon dioxide  $\text{CO}_2$  s carbon dioxide occurs naturally as a gas in the atmosphere  $\text{CO}_2$  g a which form of carbon dioxide has higher entropy b which form of carbon dioxide has higher enthalpy 2

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for waldron in seattle the running game was particularly effective using motion on offense for seattle at ninth with 4 45 yards a play caleb williams can't hit a baseball off a tee meanwhile

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