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enthalpy of solution the sum of the changes of enthalpy in each step of the formation of a solution heat of hydration enthalpy change that

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this chapter examines the factors that influence the rates of chemical reactions the mechanisms by which reactions proceed and the quantitative techniques used to describe the rates at which reactions occur

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the molar mass of an element or compound is the mass in grams of 1 mole of that substance a property expressed in units of grams per mole g mol see figure 2 5 1 figure 2 5 1 each sample contains  $6.022 \times 10^{23}$  atoms 1.00 mol of atoms from left to right top row 65.4g zinc 12.0g carbon 24.3g magnesium and 63.5g copper

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