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a quantum is the minimum amount of energy that can be lost or gained by an atom our resource for holt chemistry includes answers to chapter exercises as well as detailed information to walk you through the process step by step with expert solutions for thousands of practice problems you can take the guesswork out of studying and move forward with confidence chapter 10 electrochemical cells 10 1 1 b for the galvanic cell represented by $\text{Fe} | \text{Fe}^{2+}(\text{aq}) || \text{Ag}^{+}(\text{aq}) | \text{Ag}$ s stronger oxidizing agent silver ions stronger reducing agent iron metal unlike static pdf holt chemistry 6th edition solution manuals or printed answer keys our experts show you how to solve each problem step by step no need to wait for office hours or assignments to be graded to find out where you took a wrong turn sample answer the volume of water in cylinder a decreased to about 25 ml the volume of water in cylinder b increased to about 25 ml when the volumes approached 25 ml the increases and decreases with each transfer became smaller section 5 3 questions page 313 1 a solution step 1 use the balanced chemical equation to determine the bonding of each substance H_2 g Cl_2 g 2HCl g step 2 determine the number of moles of reactants and products the number of moles of bonds broken or formed and the molar bond energy for each general chemistry chemistry 2e openstax 22 appendices answer keys expand collapse global location the sodium and chloride ions are separated and surrounded by the water molecules because the sodium ion is positively charged it attracts the negatively charged portion of the water molecule the oxygen atom and repels the positively charged portion of the water molecule the hydrogen atoms section 10 1 assessment page 324 7 explain why chemists use the mole chemists use the mole because it is a convenient way of knowing how many representative particles are in a sample 8 state the mathematical relationship between avogadro s number and 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