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types of solution the solutions are generally classified in two ways i based on the physical state of the solution and the solute and ii based on the particle size of the solute the solutions can be classified based on the physical state of the solution solvent and solute

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in chemistry a solution is a special type of homogeneous mixture composed of two or more substances in such a mixture a solute is a substance dissolved in another substance known as a solvent

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such solvents include water methyl alcohol liquid ammonia and hydrogen fluoride the ions of the solute surrounded by dipolar molecules of the solvent are detached from each other and are free to migrate to charged electrodes such a solution can conduct electricity and the solute is called an electrolyte

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explain why certain substances dissolve in other substances the major component of a solution is called the solvent the minor component of a solution is called the solute by major and minor we mean whichever component has the greater or lesser presence by mass or by moles

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a solution is a homogeneous mixture of two substances a solute and a solvent solute substance being dissolved present in lesser amount solvent substance doing the dissolving present in larger amount

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retort alan snapped back with a witty retort see more results smart vocabulary related words and phrases solution noun mixture c or u chemistry specialized a mixture in which one substance is dissolved in another especially a solid dissolved in a liquid an aqueous solution of salts

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solutions when viewed at the molecular level have the solute particles evenly and randomly dispersed in the solvent also because the solute and solvent are in contact with each other there must be some kind of molecular interaction between the two types of molecules

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