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this page explains how to write the formula for an organic compound given its name and vice versa it covers alkanes cycloalkanes alkenes simple compounds containing halogens alcohols aldehydes and ketones an introduction to the types of formulae you will come across and the various ways of drawing organic compounds naming organic compounds understanding the code that is used to name simple organic compounds properties of organic compounds menu you can find details of the physical and chemical properties for each of these types of compound aliphatic compounds these are compounds where the functional group is not attached directly to a benzene ring alkanes alkenes halogenoalkanes haloalkanes or alkyl halides alcohols iupac nomenclature of compounds with functional groups find the longest carbon chain containing the functional group with highest priority see table 2 3 this chain determines the parent name of the compound their general formula is cnh2n 1oh the substance normally called alcohol in everyday life is just one of the family and is properly called ethanol the four smallest alcohols are structural isomerism in the alcohols this page explains how to write the formula for an organic compound given its name and vice versa it covers alkanes cycloalkanes alkenes simple compounds containing halogens alcohols aldehydes and ketones in our study of organic chemistry it will become extremely important to be able to quickly recognize the most common functional groups because they are the key structural elements that define how organic molecules react 2 2 naming organic compounds systematic names chemists communicating about organic chemistry can use drawings in many situations but in others the preference would be to describe substances with names however there are millions of known described organic substances naming organic compounds chemguide core chemistry 14 16 understanding how organic compounds are named the names of organic compounds can look quite scary when you start but there is a simple logic behind them you have to do a tiny bit of learning and then everything will fall into place as you may see from table 1 the hydroxyl group appears in two types of organic compounds alcohols and carboxylic acids while the carbonyl group is found in three aldehydes ketones acids organic chemistry studies the structure properties and reactions of organic compounds which contain carbon in covalent bonding the study of structure determines their chemical composition and formula and the study of properties includes physical and chemical properties and evaluation of chemical reactivity to understand their behavior in organic chemistry we will learn about the reactions chemists use to synthesize crazy carbon based structures as well as the analytical methods to characterize them we will also think about how those reactions are occurring on a molecular level with reaction mechanisms simply put organic chemistry is like building with molecular legos basic organic chemistry menu bonding in organic compounds includes basic electronic structure bonding in methane ethene benzene and carbonyl compounds and ideas about electronegativity and bond polarity these definitions are the preferred ones to be used in organic chemistry note that some terms such as configuration have more than one interpretation acid an agent able to produce positively charged hydrogen ions h since the hydrogen ion is a bare proton it usually exists in a solvated form such as h 3 o achiral not chiral a gical chemistry biochemistry organic chemists are also able to create or

synthesize new compounds with unique and useful pr d asa commonly called aspirin people have known since ancient times that an extract of willow bark n reduce pain and inflammation chemists eventually identified the active compound as salic organic chemistry is the chemistry of compounds of carbon carbon is unique among the other elements in that its atoms can form stable covalent bonds with each other and with atoms of other elements in a multitude of variations organic mechanisms menu free radical reactions free radical substitution reactions in alkanes and alkyl groups free radical addition during the polymerisation of ethene and the reaction between hbr and alkenes in the presence of organic peroxides electrophilic addition reactions explains how you can analyse substances using machines mass spectrometry infra red spectroscopy nmr uv visible absorption spectrometry and chromatography basic organic chemistry includes help on bonding naming and isomerism and a discussion of organic acids and bases properties of organic compounds the simplest class of organic compounds is the hydrocarbons which consist entirely of carbon and hydrogen petroleum and natural gas are complex naturally occurring mixtures of many different hydrocarbons that furnish raw materials for the chemical industry organic chemistry is the chemistry of carbon based compounds other than the most simple ones like carbon dioxide and carbonates when you start organic chemistry it can be quite confusing on this page we are just going to look at the various ways organic formulae can be written and organic structures drawn

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iupac nomenclature of compounds with functional groups find the longest carbon chain containing the functional group with highest priority see table 2 3 this chain determines the parent name of the compound

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their general formula is cnh2n 1oh the substance normally called alcohol in everyday life is just one of the family and is properly called ethanol the four smallest alcohols are structural isomerism in the alcohols

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in our study of organic chemistry it will become extremely important to be able to quickly recognize the most common functional groups because they are the key structural elements that define how organic molecules react

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as you may see from table 1 the hydroxyl group appears in two types of organic compounds alcohols and carboxylic acids while the carbonyl group is found in three aldehydes ketones acids

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organic chemistry studies the structure properties and reactions of organic compounds which contain carbon in covalent bonding the study of structure determines their chemical composition and formula and the study of properties includes physical and chemical properties and evaluation of chemical reactivity to understand their behavior

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in organic chemistry we will learn about the reactions chemists use to synthesize crazy carbon based structures as well as the analytical methods to characterize them we will also think about how those reactions are occurring on a molecular level with reaction mechanisms simply put organic chemistry is like building with molecular legos

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the simplest class of organic compounds is the hydrocarbons which consist entirely of carbon and hydrogen petroleum and natural gas are complex naturally occurring mixtures of many different hydrocarbons that furnish raw materials for the chemical industry

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