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[supersaturated solution definition and examples science](#) Apr 11 2024 2022 by definition a supersaturated solution is a chemical solution that contains more solute than the solvent can hold in other words a supersaturated solution has more dissolved solute than a saturated solution the process of forming a supersaturated solution is called supersaturation

[16 5 supersaturated solutions chemistry libretxts](#) Mar 10 2024 2022 a supersaturated solution is a solution that contains more than the maximum amount of solute that is capable of being dissolved at a given temperature the recrystallization of the excess dissolved solute in a supersaturated solution can be initiated by the addition of a tiny crystal of solute called a seed crystal

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[supersaturation wikipedia](#) Jan 08 2024 in physical chemistry supersaturation occurs with a solution when the concentration of a solute exceeds the concentration specified by the value of solubility at equilibrium most commonly the term is applied to a solution of a solid in a liquid but it can also be applied to liquids and gases dissolved in a liquid

[7 10 solubility saturated unsaturated and supersaturated](#) Dec 07 2023 2024 because the amount of solute that is contained in a supersaturated solution exceeds the natural quantity of solute that should be dissolved supersaturated solutions are highly unstable and the excess solute readily crystallizes if

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[10 16 saturated and supersaturated solutions chemistry](#) Oct 05 2023 2023 such a solution is said to be supersaturated a good example of supersaturation is provided by sodium thiosulfate whose solubility at 25 c is 50 g per 100 g of water if 70 g of sodium thiosulfate crystals is dissolved in 100 g of hot water and the solution cooled to room temperature the extra 20 g of sodium thiosulfate

[saturated and supersaturated solutions chemistry socratic](#) Sep 04 2023 key questions what is a supersaturated solution a supersaturated solution contains more solute at a given temperature than is needed to form a saturated solution increased temperature usually increases the solubility of solids in liquids for example the solubility of glucose at 25 c is 91 g 100 ml of water

[supersaturation and crystallization harvard natural](#) Aug 03 2023 a supersaturated solution is unstable and by seeding it you can trigger rapid crystallization how it works sodium acetate can dissolve in water in great quantities at high temperature and if you let the solution cool carefully to

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