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a element is one in which the nucleus breaks down and gives off particles and energy radioactive elements in groups 3 through 12 on the periodic table are known as elements transition the two rows of elements that seem to be disconnected from the rest of the periodic table are the

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what are common properties of metals answer metals are good conductors of heat and electricity reflect light are malleable and ductile and except for mercury are solid at room temperature section check

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elements are generally divided into two main groups metals and nonmetals both groups have a number of subgroups all elements have

certain properties for example most metals are hard shiny solids they are good conductors of both heat and electricity metals are also malleable and ductile

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review quadrilaterals their properties find the measurement indicated in each parallelogram 1 f e d c 123 2 d c e f 3 w v u t 110 solve for x each figure is a parallelogram 4 r s u t 60 70 13 x 2 5 s r t u x 66 70 x 44 1

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conceptual answers the 1 s shell is closer to the nucleus and therefore experiences a greater electrostatic attraction in addition the electrons in the 2 s subshell are shielded by the filled 1 s 2 shell which further decreases the electrostatic attraction to the nucleus

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in this article we will explore the key aspects of elements and their properties providing answers to commonly asked questions the periodic table is a powerful tool in understanding the elements and their properties

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how many are metalloids how many are nonmetals consult a periodic table to answer 2 combined atoms and uncombined atoms behave differently this is due to the difference in their electron arrangement consider these four species determine the number of protons neutrons and electrons in each

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reason 1 location is very important for example i think a moderate property in a good neighborhood is better than really good property in a bad neighborhood also access to public transportation is very important because i don t want to drive all the time reason 2 i also look at the condition of the property such as when it was built

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