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11 3 solubility chemistry 2e openstax

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the temperature dependence of solubility can be exploited to prepare supersaturated solutions of certain compounds a solution may be saturated with the compound at an elevated temperature where the solute is more soluble and subsequently cooled to a lower temperature without precipitating the solute

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salt dissolved in water is a solution the major component of a solution called the solvent is typically the same phase as the solution itself each minor component of a solution and there may be more than one is called the solute

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solution in chemistry a homogenous mixture of two or more substances in relative amounts that can be varied continuously up to what is called the limit of solubility the term solution is commonly applied to the liquid state of matter but solutions of gases and solids are possible

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solutions are composed of particles the size of an ion or small molecule 0.1 to 200 nm the examples provided above are all considered solutions air is a solution of small gas molecules simple syrup is a solution of sucrose in water and seawater is a solution of ions and water

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13 1 types of solutions some terminology in all solutions whether gaseous liquid or solid the substance present in the greatest amount is the solvent and the substance or substances present in lesser amounts are the solutes

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a solution is a homogeneous mixture of one or more solutes dissolved in a solvent note that the solvent is the substance that is present in the greatest amount many different kinds of solutions exist for example a solute can be a gas a liquid or a solid solvents can

also be gases liquids or solids

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the meaning of solution is an action or process of solving a problem how to use solution in a sentence

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updated on october 24 2022 a solution is a homogeneous mixture of two or more substances a solution may exist in any phase a solution consists of a solute and a solvent the solute is the substance that is dissolved in the solvent the amount of solute that can be dissolved in solvent is called its solubility

solution chemistry wikipedia

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in chemistry a solution is a special type of homogeneous mixture composed of two or more substances in such a mixture a solute is a substance dissolved in another substance known as a solvent

12 3 types of solutions and solubility chemistry fundamentals

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a solution that contains a relatively low concentration of solute is called dilute and one with a relatively high concentration is called concentrated if we add more salt to a saturated solution of salt we see it fall to the bottom and no more seems to dissolve

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types of solutions a solution is defined as a homogenous mixture comprising of two or more components of solute and solvent know more about mixtures and solutions with examples at byju s

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3 answers sorted by 69 i need to find a solution to this problem i need to find a solution for this problem i need to find a remedy for this problem all three sentences are correct although the second is less common usually one will find that to is the preposition of choice for the word solution

solution noun definition pictures pronunciation and usage

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countable a way of solving a problem or dealing with a difficult situation synonym answer attempts to find a solution have failed to propose offer provide develop a solution to look for seek a solution it was decided that the only solution was for him to leave the company an easy innovative solution a possible practical long term solution

5 1 introduction to solution chemistry libretexts

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what is a solution a solution is a homogeneous mixture of two or more pure substances the substance that is in a large amount in the solution is called the solvent the substance that is in smaller amounts in a solution is called the solute

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noun us sə'luʃən solution noun answer add to word list c an answer to a problem it seemed a reasonable solution to a difficult problem

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a solution in science is a homogenous mixture of two or more substances solutions appear to be one substance but the parts of a solution are not chemically bonded solutions can

exist in any phase of matter and the proportions of substances in a solution can vary up to the limit of solubility

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