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□□□ 2024□3□26□ air for example is a solution consisting chiefly of oxygen and nitrogen with trace amounts of several other gases and brass is a solution composed of copper and zinc a brief treatment of solutions follows for full treatment see liquid solutions and solubilities life processes depend in large part on solutions □□□ 2023□6□5□ solution is a homogeneous mixture of two or more components in which the particle size is smaller than 1 nm common examples of solutions are sugar in water and salt in water solutions soda water etc in a solution all the components appear as a single phase □□□ example make 2 g 100ml of nacl solution with 1 l water the density of the resulting solution is considered to be equal to that of water statement holding especially for dilute solutions so the density information is not required □□□ solution example□□□ □□□ □□□ on the web solution example property solution example property and a property solution example property and a solution example property a solution example _____ __ __ 1 example solution ___ __ __ n example solution non nonno nonnonnonno nonnonnon non $2023\Pi 4\Pi 3\Pi$ for example the air is gas in a gas solution carbonated water is a gas in a liquid solution vinegar is a liquid in a liquid solution metal alloys are solid in solid solutions table 5 1 1 lists the major types of solutions solvents and primary solutes in □□□ 2022□7□19□ solutions exist for every possible phase of the solute and the solvent salt water for example is a solution of solid ce nacl in liquid water while air is a solution of a gaseous solute ce o2 in a gaseous solvent ce ∏∏∏ 2023∏1∏12∏ solution examples examples of solutions include air sugar water steel saltwater pancake syrup and natural gas examples of solutions air is an example of a gaseous solution gas gas the air we breathe exists in roughly n □□□ 2023□3□24□ a solution is what occurs when two chemicals are mixed referred to as a solvent and a solute the combination of solute and solvent together is a solution these are often one substance dissolved in another such as dissolving salt in water although a solution can be made from any state of matter combining a stoolneths iandanal solvent 1 2022 10 24 solution examples any timovestigative accounting 5th edition

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substances which can be evenly mixed may form a solution even though materials of different phases may combine to form a solution the end result always exists of a single phase an example of a solid solution is brass an example of a liquid solution is aqueous hydrochloric acid hcl in water [][] 2020[4] [19] examples of solutions the solute and solvent may be the same phase as one another e g gases in air or in different phases e g dissolving salt in water to make saline solution the end result is always a single phase here are some examples of solutions in different phases

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